

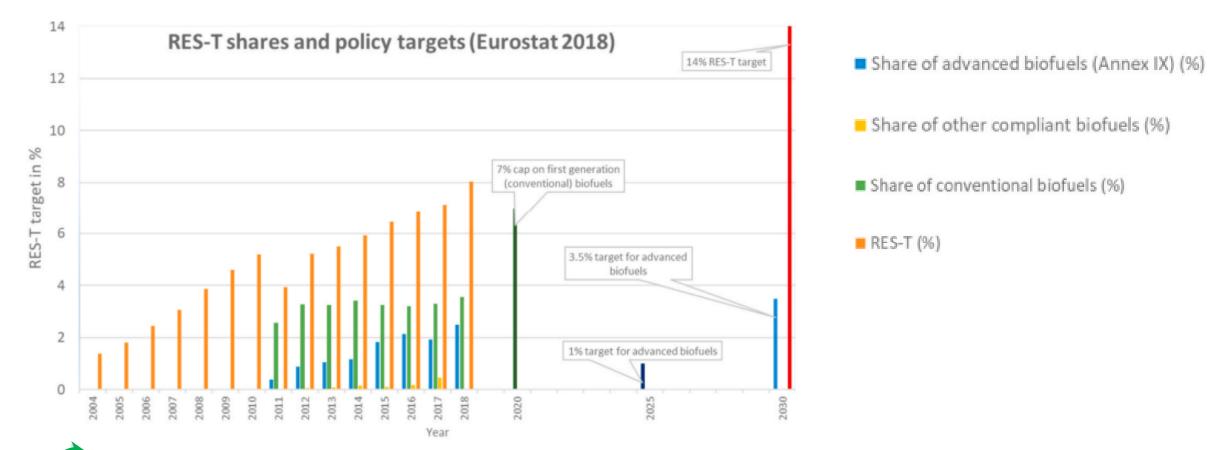
Membrane reactor to enhance a methanol production from CO₂ and H₂ in biomass to biodiesel route

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18th May 2022, Innovations in advanced biofuels production, Converge workshop & TNO lab tour

Background

- Advanced biofuels are essential for the transition to zero carbon
- Production costs are higher than their fossil counterparts: significant innovation, technological development and scale up needed

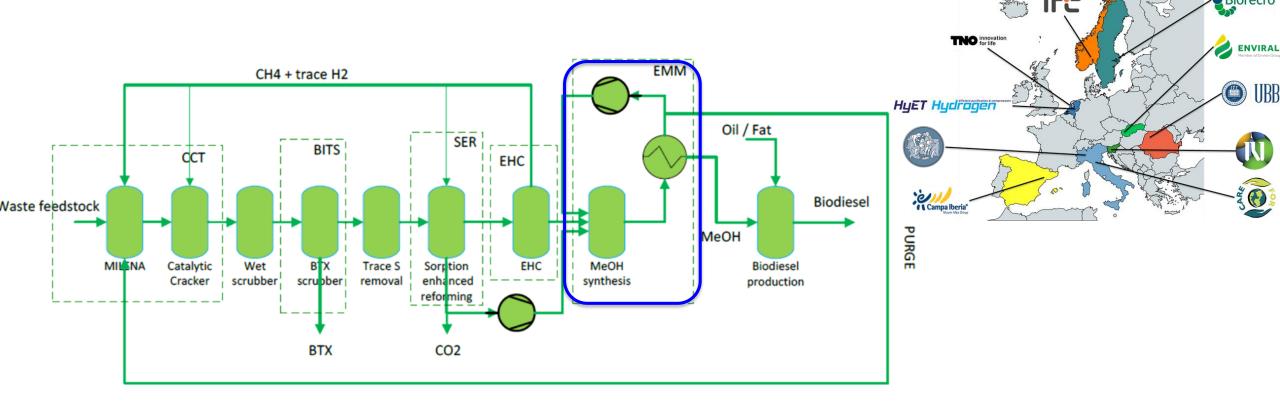


Objectives: Converge project

 The CONVERGE project aims to increase efficiency of the biodiesel production by 12% per secondary biomass unit used, and reduce the CAPEX by 10%

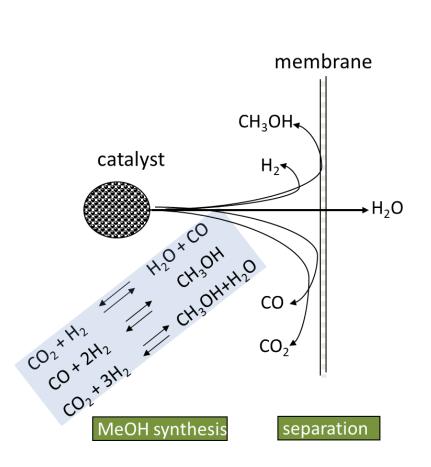
The CONVERGE technologies will be validated for more than 2000 cumulated hours

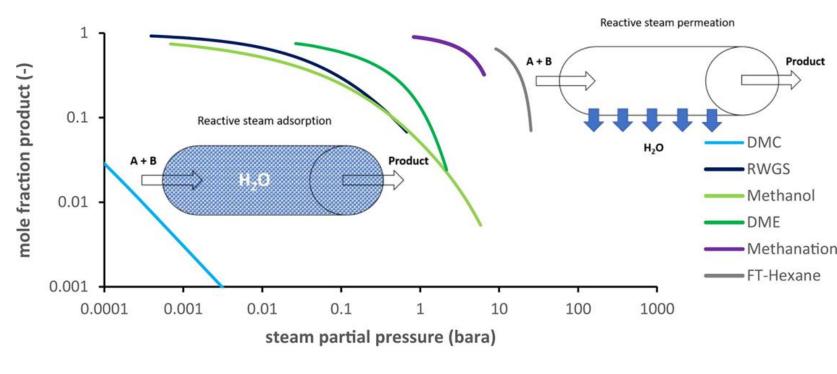
taking these from the TRL3 to development stage TRL5.





Separation enhanced reactions



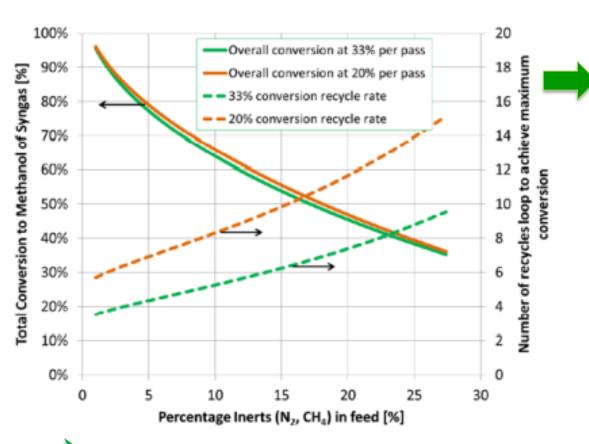


Significant increase of MeOH yield at moderate steam partial pressures



Objectives: membrane assisted methanol production

- Develop stable membranes at reaction conditions
- Develop multi-tube membrane reactor, targeted conversion for feed CO₂/H₂ 33% per pass
- Demonstration of integrated process at TRL 5



Decrease of recycle:

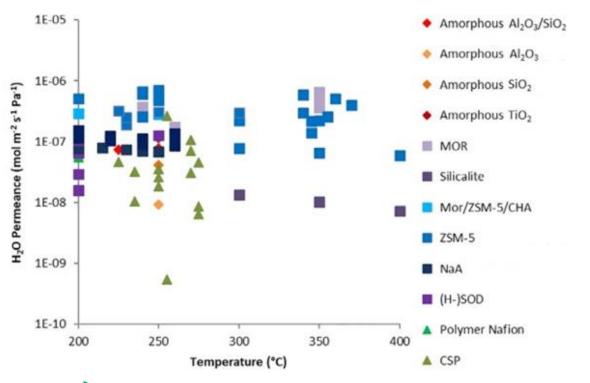
- Decrease of reactor size
- Reduction in energy penalty of the recycle compressors

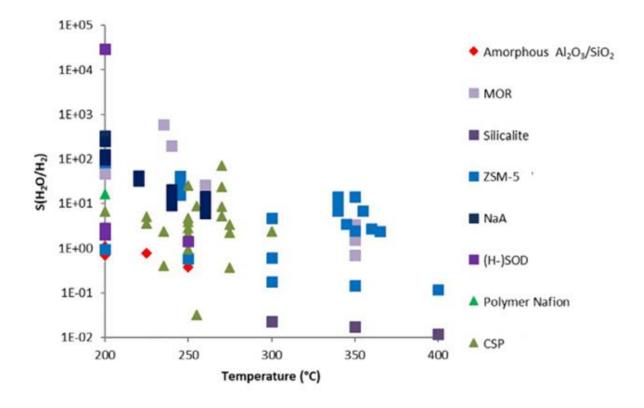


Membrane development -target

Membrane development targets:

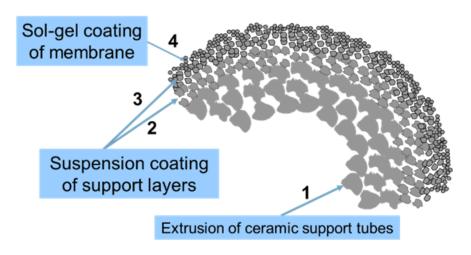
- 1) Stability at the methanol operating T and p (T = 220-275 $^{\circ}$ C, p up to 100 bar)
- 2) High selectivity for steam and methanol $CO_2 + 3H_2 = CH_3OH + H_2O$
- 3) High steam/methanol permeability 👈 high flux



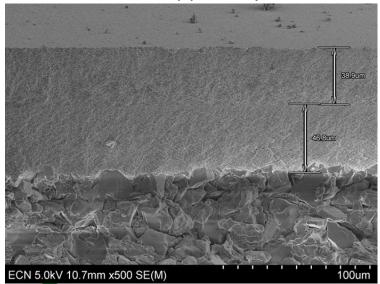


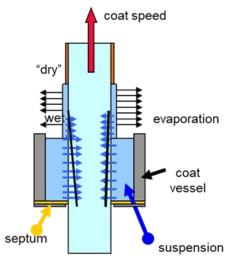


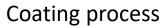
Membrane synthesis procedure



Membrane support layers

















CarbON Valorisation in Energy-efficient Green fuels

Membrane selection

Amorphous microporous

APTES-PA (Aminopropyl triethoxysilane-Polyamide)

BTESE (1, 2-
$$\underline{b}$$
is (\underline{t} ri \underline{e} thoxy \underline{s} ilyl) \underline{e} thane) $\underline{h_3C}$ \underbrace{o} \underbrace{o} \underbrace{o} \underbrace{o} \underbrace{c} $\underbrace{h_3C}$ \underbrace{o} \underbrace{o} \underbrace{c} $\underbrace{h_3C}$ \underbrace{o} \underbrace{o} \underbrace{c} \underbrace{c} $\underbrace{h_3C}$ \underbrace{o} \underbrace{c} \underbrace{c} \underbrace{c} $\underbrace{h_3C}$ \underbrace{o} \underbrace{c} \underbrace{c} \underbrace{c} \underbrace{d} \underbrace{c} \underbrace{d} \underbrace{c} \underbrace{d} $\underbrace{d$

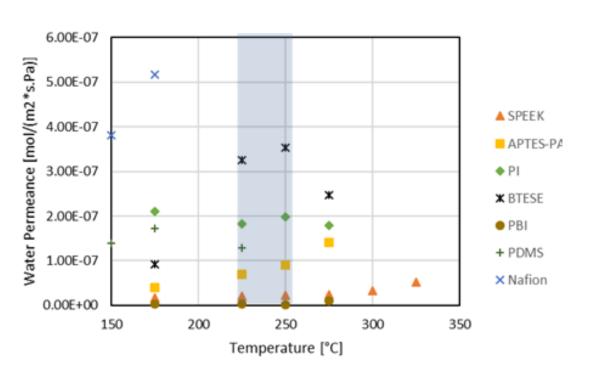
Polymeric

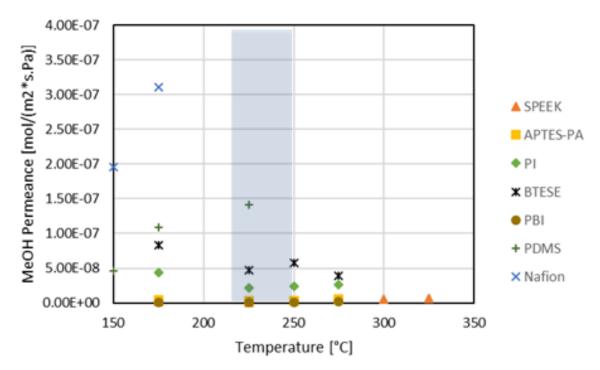
SPEEK (sulfonated poly(ether ether ketone))

Membrane separation test results

Test conditions:

- $p_{feed} = 35 \text{ bar, } p_{perm} = 1.5 \text{ bar, no sweep}$
- 60% H₂, 10% (50/50)methanol/steam, 20% CO₂, 1% CO, 9% N₂





- Nafion, BTESE, PI highest steam and MeOH permeance
- BTESE performance decreases at 275°C, Nafion not selective at T>200°C, PDMS not selective T > 225°C
- SPEEK and PBI low H₂O and MeOH permeance



Conclusions: Membrane characterisation

PI membrane preselected as the most promising to reach conversion targets. (T_{range} = 225-250°C)

• H₂O/H₂ selectivity:

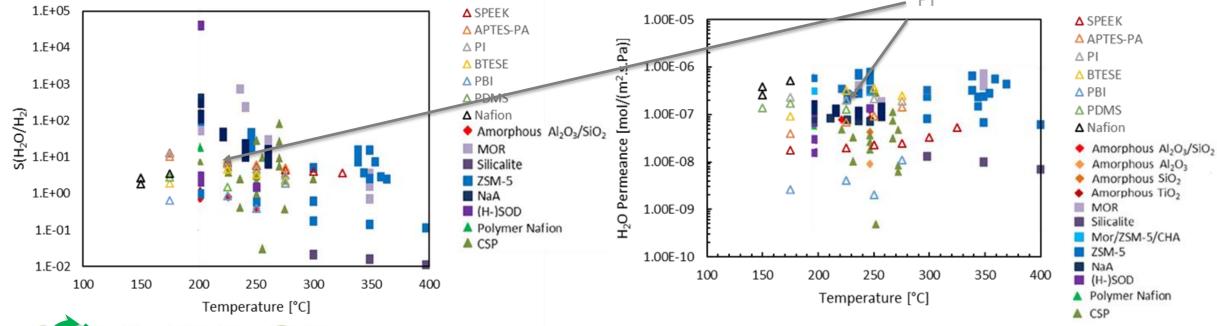
• MEOH/H₂ selectivity:

H₂O permeance:

MeOH permeance:
H₂O>H₂>MEOH>CO₂>CO≈N₂

PI	BTESE	APTES-PA	
4.7-6.5	3.5-4.3	6-8	
0.6-0.8	0.6-0.7	0.2-0.4	
PI	1.6 [.] PI	PI/2.3	
PI	2.2 [.] PI	PI/8.4	

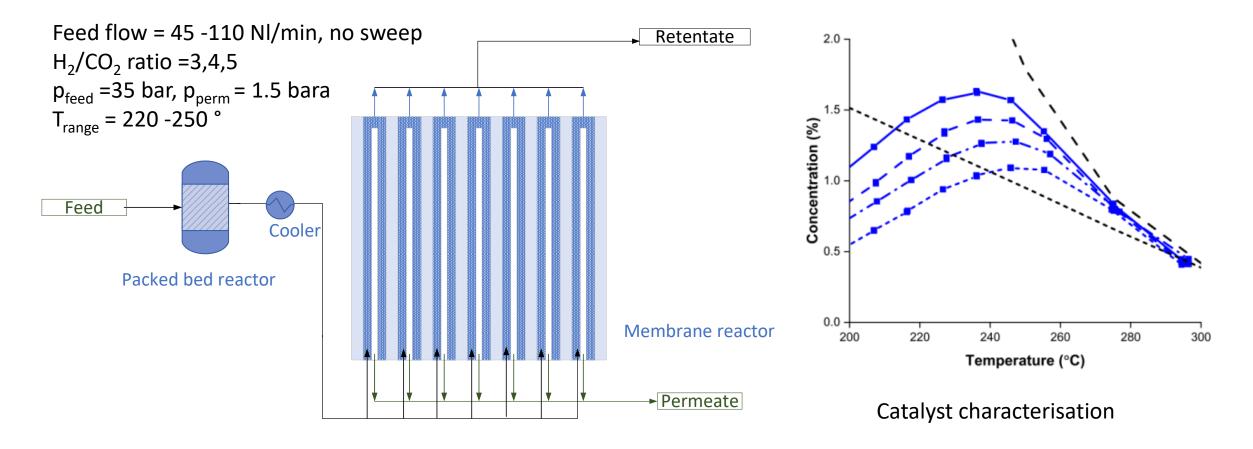
Steam/H₂ behaviour compares well to literature





Multi-tubular membrane reactor

- Multi-tubular membrane reactor constructed with 7 PI membranes of 80 cm effective length, $A_{mem} = 0.25 m^2$
- Commercial MeOH catalyst (ALFA AESAR)





Multi-tubular membrane reactor construction











Multi-tubular membrane reactor

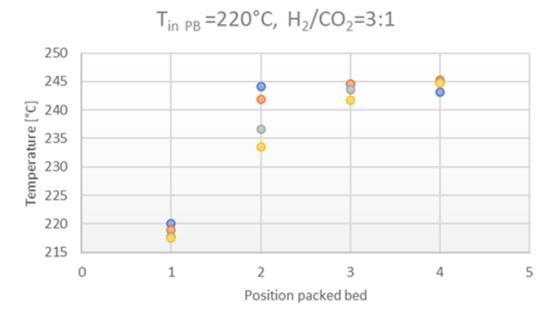
Membranes installed in the membrane reactor





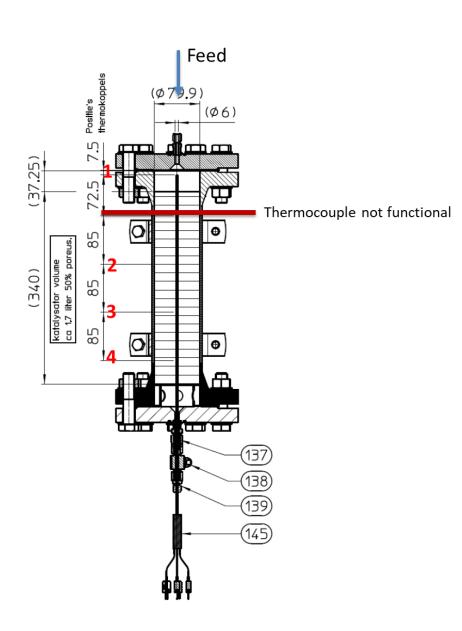
Packed bed reactor results

- WHSV 0.60-1.32 h⁻¹
- 5 thermocouples in the packed bed (one not functional)
- Equilibrium conversion reached for all WHSV, and for all tested T



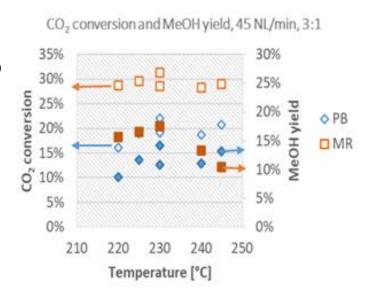
● Flow= 45 Nl/min ● Flow = 65 NL/min ● Flow = 85 NL/min ● Flow=105NL/min

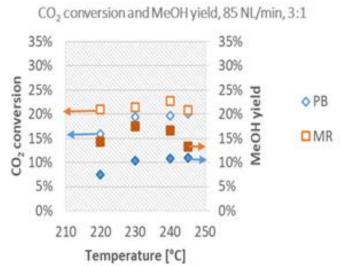


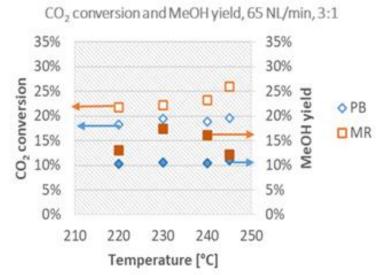


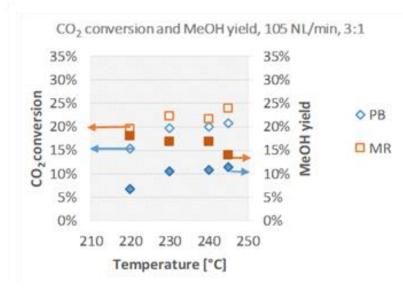
Multi tubular membrane reactor, stochiometric conditions

- WHSV = 0.2- 0.4 h⁻¹
- MR CO₂ and MeOH yield increased compared to packed bed reactor
- Highest increase observed for the lowest feed flow corresponding to WHSV = $0.2 h^{-1}$ at 230 °C:
 - → 30 % vs 22% MR vs PB CO₂ conversion
 - → 18% vs. 11% MR vs PB MeOH yield



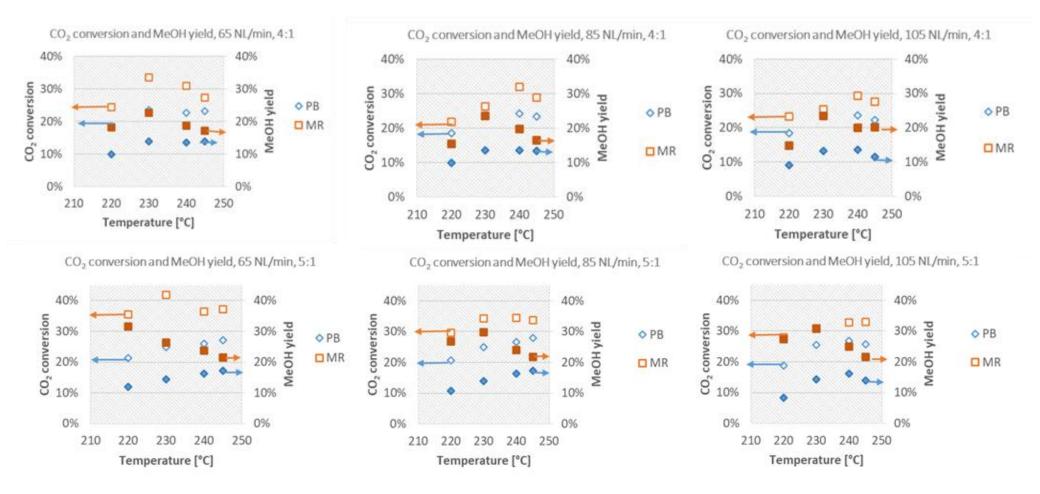








Multi tubular membrane reactor, H₂ excess



• As expected, with increase of H_2 in the feed, overall CO_2 conversion and MeOH yield increase. Max. achieved 65 NL/min, H_2/CO_2 =5, T = 230°C : \rightarrow 42 % vs 25% MR vs PB CO_2 conv., 26% vs. 14% MR vs PB MeOH YLD



BBBBBB Temperature profiles per membrane at different feed flows DETAIL2 Membrane T profile - feed flow 65 NL/min,3 1 Membrane T profile -feed flow 45 NL/min, Membrane T profile - feed flow 85 NL/min, 3:1 240 240 240 Ç 235 Ç 235 ♦ Mem1 **୍ଟ୍ର** 235 Temberature 225 220 215 230 230 225 □ Mem2 225 △ Mem3 220 215 430 - 767.5 220 ×Mem4 **X** Mem5 210 210 OMem6 210 240 - 577.5 + Mem7 Position membrane tube Position membrane tube Position membrane tube Membrane T profile - feed flow 105 NL/min 240 235 ♦ Mem1 ■ Mem2 △ Mem3 DETAIL1 ed 225 ×Mem4 Feed entrance x Mem5 220 o Mem6 Doorsnede A-A +Mem7 215 Position membrane tube



- Thermocouples are positioned in middle of the membrane tube

Conclusions

- 5 polymeric and 2 hybrid silica membranes characterized on separation performance
- Membrane performance comparable to reported in the literature
- PI membrane had the best performance → installed in multi-tubular membrane reactor
- Multi-tubular membrane reactor with $A_{mem} = 0.25 \text{ m}^2$ constructed, MeOH production capacity 0.5L/h
- MR CO₂ and MeOH yield increased compared to packed bed reactor
- Highest increase for stochiometric composition observed for the lowest feed flow corresponding to WHSV = $0.2 h^{-1}$ at 230 °C:
 - → 30 % vs 22% MR vs PB CO₂ conversion
 - → 18% vs. 11% MR vs PB MeOH yield
- As expected, in H₂ excess, an overall CO₂ conversion and MeOH yield increase
- Maldistribution of the flow observed in the membrane reactor at low feed flows





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CONVERGE: CarbON Valorisation in Energy-efficient Green fuels

Questions?

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Long term membrane performance

- Experimental campaign lasted 3168 h, with the reactive tests running for 1896 h
- Between 1274-1776 h the activation of the catalyst took place → performance of the membranes was not affected by catalyst activation.
- The increase of the N₂ flux was observed at T of 240 °C at 2664h, after emergency shut down in which membrane reactor system cooled down rapidly
- N₂ flux at 185 °C at the end of the experimental campaign was therefore increased for 22% compared to the measured before catalyst activation

